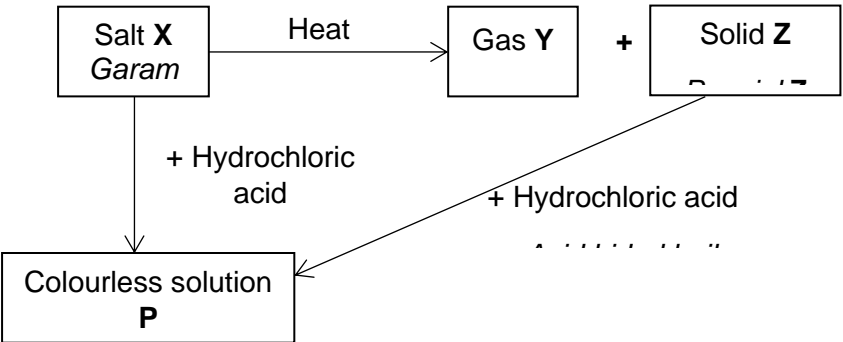
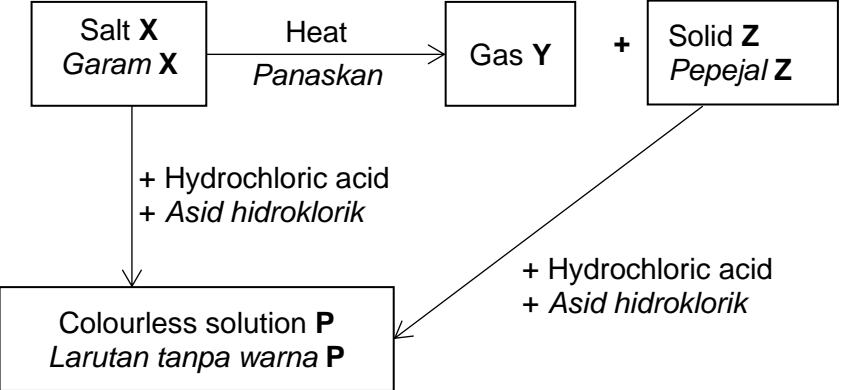
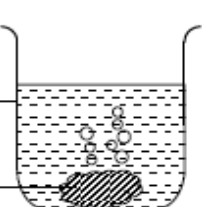
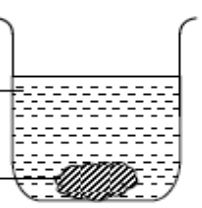
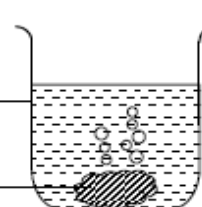
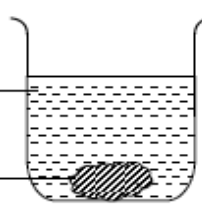
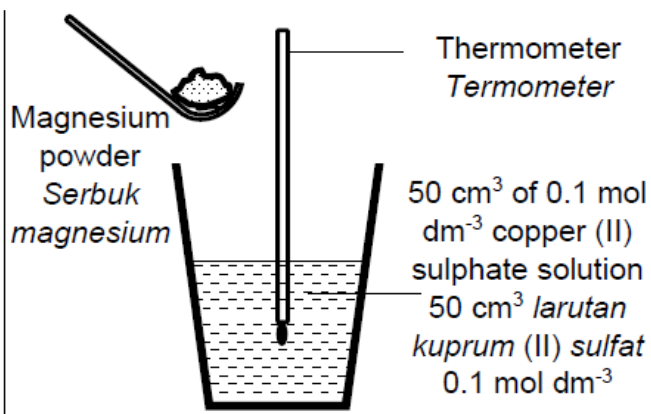
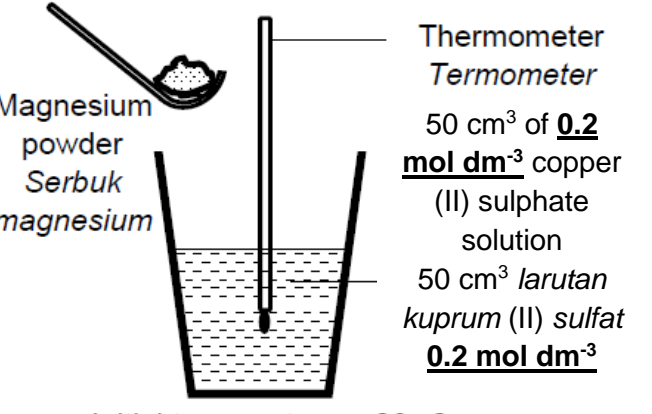


ERRATA UJIAN DIAGNOSTIK (2)
KIMIA 2 (4541/2)

M/S	NO SOALAN	KANDUNGAN ASAL	PEMBETULAN
18	(c)	 <pre> graph TD A["Salt X Garam"] -- Heat --> B["Gas Y"] A -- Heat --> C["Solid Z"] A -- "+ Hydrochloric acid" --> D["Colourless solution P"] C -- "+ Hydrochloric acid" --> D </pre>	 <pre> graph TD A["Salt X Garam X"] -- Heat Panaskan --> B["Gas Y"] A -- Heat Panaskan --> C["Solid Z Pepejal Z"] A -- "+ Hydrochloric acid + Asid hidroklorik" --> D["Colourless solution P Larutan tanpa warna P"] C -- "+ Hydrochloric acid + Asid hidroklorik" --> D </pre> <p><i>**Dengan terjemahan Bahasa Melayu</i></p>
	(c)	<p>... Solid Z is a <u>brown solid that turns yellow</u> when cooled....</p> <p>... <i>Pepejal Z ialah pepejal berwarna perang yang bertukar menjadi kuning setelah sejuk...</i></p>	<p>...Solid Z is a <u>yellow solid that turns white</u> when cooled...</p> <p>...<i>Pepejal Z ialah pepejal kuning yang bertukar putih setelah sejuk...</i></p> <p><i>** Tukarkan ayat didalam soalan</i></p>

M/S	NO SOALAN	KANDUNGAN ASAL	PEMBETULAN
20	7 (c)	<p>II</p> <p>Glacial CH_3COOH in solvent K <i>CH₃COOH glasial dalam pelarut K</i></p> <p>Calcium carbonate <i>Kalsium karbonat</i></p>  <p>Glacial CH_3COOH in solvent K <i>CH₃COOH glasial dalam pelarut K</i></p> <p>Calcium carbonate <i>Kalsium karbonat</i></p> 	<p>II</p> <p>Glacial CH_3COOH in solvent K <i>CH₃COOH glasial dalam pelarut K</i></p> <p>Calcium carbonate <i>Kalsium karbonat</i></p>  <p>Glacial CH_3COOH in solvent L <i>CH₃COOH glasial dalam <u>pelarut L</u></i></p> <p>Calcium carbonate <i>Kalsium karbonat</i></p> 
			**Tukarkan pelarut K kepada pelarut L bagi bikar kedua Set II
23	8 (c)	<p>II</p> <p>Magnesium powder <i>Serbuk magnesium</i></p> <p>Thermometer <i>Termometer</i></p> <p>50 cm³ of 0.1 mol dm⁻³ copper (II) sulphate solution <i>50 cm³ larutan kuprum (II) sulfat 0.1 mol dm⁻³</i></p> <p>Initial temperature = 28 °C <i>Suhu awal</i></p> 	<p>II</p> <p>Magnesium powder <i>Serbuk magnesium</i></p> <p>Thermometer <i>Termometer</i></p> <p>50 cm³ of 0.2 mol dm⁻³ copper (II) sulphate solution <i>50 cm³ larutan kuprum (II) sulfat <u>0.2 mol dm⁻³</u></i></p> <p>Initial temperature = 28 °C <i>Suhu awal</i></p> 
			**Tukarkan kepekatan kuprum (II) sulfat kepada 0.2 mol dm ⁻³